

BiScO₃: Centrosymmetric BiMnO₃-type Oxide

Alexei A. Belik,^{*,†} Satoshi Iikubo,[‡] Katsuaki Kodama,[‡] Naoki Igawa,[‡] Shin-ichi Shamoto,[‡] Makoto Maie,[§] Takuro Nagai,[¶] Yoshio Matsui,[§] Sergey Yu. Stefanovich,[#] Bogdan I. Lazoryak,[#] and Eiji Takayama-Muromachi[§]

International Center for Young Scientists (ICYS), High Voltage Electron Microscopy Station (HVEMS), and Advanced Materials Laboratory (AML), National Institute for Materials Science (NIMS), 1-1 Namiki, Tsukuba, Ibaraki 305-0044, Japan, Neutron Science Research Center, Japan Atomic Energy Research Institute (JAERI), Tokai, Ibaraki 319-1195, Japan, and Department of Chemistry, Moscow State University, Moscow 119992, Russia

Received November 6, 2005; E-mail: Alexei.Belik@nims.go.jp

Bi- and Pb-containing perovskites with transition metals have received much attention in recent years due to the renewed great interest in multiferroic materials.^{1–4} In multiferroic systems, two or all three of (anti)ferroelectricity, (anti)ferromagnetism, and ferroelasticity occur in the same phase. Such systems are rare in nature⁵ but have potentially wide technological applications.^{1–4} The most attractive possible application is in multiple state memory elements.²

BiMnO₃^{6–12} and BiFeO₃¹³ are the most studied multiferroic systems among BiMO₃ with magnetic M ions. BiMnO₃ is believed to crystallize in the noncentrosymmetric space group *C2* to fulfill the requirement of ferroelectricity.^{10,11} This structural model is generally accepted for BiMnO₃ and related compounds, BiCrO₃¹⁴ and Bi₂MnNiO₆.³ Despite some efforts to understand the origin of ferroelectricity in BiMnO₃,^{6,7} it is not yet completely understood.^{8,12} On the other hand, little is known about simple Bi-containing compounds BiM'O₃ with nonmagnetic M' ions (M' = Al, Sc, Ga, and In). Recent theoretical studies of BiAlO₃ and BiGaO₃ predicted that they should be high-performance piezoelectrics and ferroelectrics with very large spontaneous polarization.¹⁵ Solid solutions including BiM'O₃ as the end members, for example, PbTiO₃–BiScO₃ and PbTiO₃–BiGaO₃,¹⁶ were extensively studied recently in the attempt to improve ferroelectric properties of PbTiO₃ and to reduce the amount of lead. In addition, BiM'O₃ can be model systems for theoretical understanding of the origin of ferroelectricity in multiferroic BiMO₃ because no additional complication of magnetic behavior exists in BiM'O₃.¹⁵

BiScO₃^{16,17} was prepared at 6 GPa in a belt-type high-pressure apparatus at 1413 K for 40 min. Neutron powder diffraction (NPD) data of BiScO₃ were collected at room temperature with a high-resolution powder diffractometer (the JRR-3M reactor in JAERI, Tokai, $\lambda = 1.8233(10)$ Å). Diffraction data were analyzed by the Rietveld method with RIETAN-2000. Indexing Bragg reflections in the X-ray powder diffraction and NPD of BiScO₃ revealed it to crystallize in the monoclinic system with lattice parameters of $a = 9.8899(5)$ Å, $b = 5.8221(3)$ Å, $c = 10.0462(5)$ Å, and $\beta = 108.300(3)^\circ$. These lattice parameters are similar to those of BiMnO₃ ($a = 9.5323$ Å, $b = 5.6064$ Å, $c = 9.8535$ Å, and $\beta = 110.667^\circ$),¹⁰ BiCrO₃,¹⁴ and Bi₂MnNiO₆,³ indicating that BiScO₃ should have a very similar structure. Reflection conditions derived from the indexed reflections of BiScO₃ were $h + k = 2n$ for hkl , and $h = 2n$ and $l = 2n$ for $h0l$, affording possible space groups *Cc* (No. 9) and *C2/c* (No. 15). The reflection conditions were checked and confirmed using electron diffraction. No reflections with $l = 2n+1$

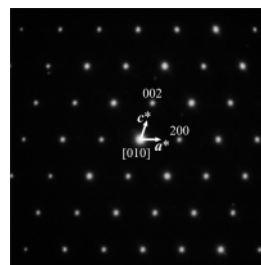


Figure 1. Electron diffraction pattern of BiScO₃ along the [010] zone axis.

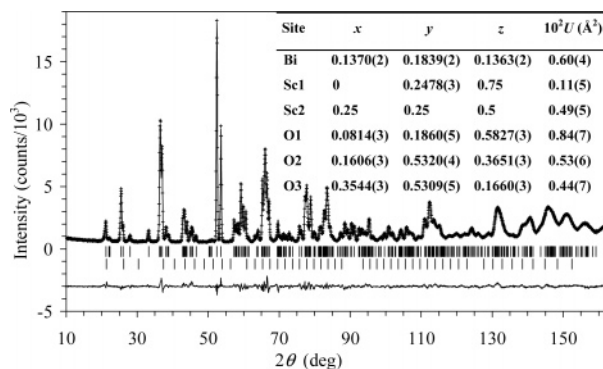


Figure 2. Observed (crosses), calculated (solid line), and difference patterns resulting from the Rietveld analysis of the neutron powder diffraction data for BiScO₃ at room temperature in space group *C2/c*. Bragg reflections are indicated by tick marks. The lower tick marks are given for the impurity phase, Sc₂O₃ (0.3 mass %). Final fractional coordinates and thermal parameters are listed.

were observed for $h0l$ and $00l$ (Figure 1). These data unambiguously confirmed that the space group of BiScO₃ cannot be *C2* (or *Cm* or *C2/m*) such as in BiMnO₃.¹⁰ No signal of second-harmonic generation was observed for BiScO₃, which shows it to be centrosymmetric. All these facts leave one possibility for BiScO₃, that is, space group *C2/c*.

Structure parameters of BiScO₃ were successfully refined from NPD in space group *C2/c* using the fractional coordinates of BiMnO₃¹⁰ transformed to space group *C2/c* as the initial ones. *R* factors were $R_{wp} = 5.38\%$ ($S = 2.07$), $R_p = 3.98\%$, $R_B = 1.70\%$, and $R_F = 0.83\%$. Thermal parameters for all the sites were typical and positive, and estimated standard deviations (esd's) of fractional atomic coordinates and thermal parameters were small (for example, $3\text{--}5 \times 10^{-4}$ for oxygen coordinates and 7×10^{-2} for oxygen thermal parameters; Figure 2). Figure 2 displays observed, calculated, and difference NPD patterns for BiScO₃.

The bond valence sum calculations resulted in the values of 2.89 for Bi, 2.91 for Sc1, and 2.92 for Sc2, confirming the formal

[†] ICYS.

[‡] JAERI.

[§] AML.

[¶] HVEMS.

[#] Moscow State University.

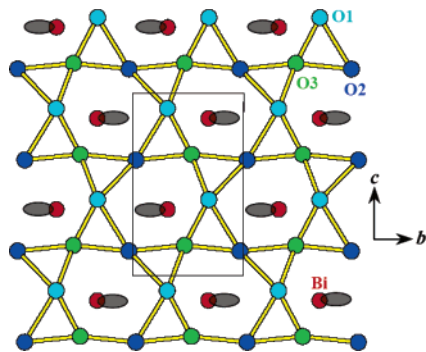


Figure 3. Projection of the structure of BiScO₃ along the *a* axis. One layer of O atoms (with bonds) and Bi atoms (with lone electron pair) is shown. Thin lines give a unit cell.

oxidation state +3 for Bi and Sc ions. Three Bi–O distances were short (2.15–2.25 Å), and other Bi–O distances were long (2.55–3.26 Å). Therefore, the lone electron pair of a Bi³⁺ ion is active in BiScO₃ (Figure 3), resulting in the antiparallel displacement of Bi³⁺ ions in the *b* direction from the centroids of their O coordination environments in the same manner as in BiMnO₃ with the *C2* model.

We also tried to apply the *C2* model as in the case of BiMnO₃, BiCrO₃, and Bi₂NiMnO₆ using the atomic coordinates of BiMnO₃¹⁰ as the initial ones in the structure analysis of BiScO₃. The *R* factors in the *C2* model ($R_{wp} = 5.14\%$ ($S = 1.99$), $R_p = 3.82\%$, $R_B = 1.50\%$, and $R_F = 0.72\%$) were slightly smaller than those of the *C2/c* model because of the larger number of the refined structural parameters (39 vs 19). However, thermal parameters of two O atoms were negative. The esd's of fractional atomic coordinates (for example, $8\text{--}24 \times 10^{-4}$ for O) and thermal parameters (for example, 0.2–0.3 for O) of all the atoms were about 1 order of magnitude larger than one may expect from NPD data.

The esd's of fractional atomic coordinates and thermal parameters in the *Cc* model (the number of refined structural parameters was 38) were almost the same as those of the *C2* model with slightly smaller *R* factors in the *Cc* model: $R_{wp} = 5.02\%$ ($S = 1.94$), $R_p = 3.70\%$, $R_B = 1.33\%$, and $R_F = 0.56\%$. Two O atoms also had negative thermal parameters in the *Cc* model. The negative thermal parameters for O atoms and large esd's for all the sites obtained from NPD data are good indicators that the symmetry of BiScO₃ is higher than *C2* or *Cc* and a center of symmetry is missed.¹⁸ It is known that disregarding a center of symmetry results in errors in the atomic coordinates that are large compared with normal esd's.¹⁸

Checking the available structural data for BiMnO₃ and related compounds revealed the same problem, which was systematically ignored, with the *C2* model. In the first paper reporting the crystal structure of BiMnO₃ from NPD at room temperature, esd's for O coordinates were $1\text{--}5 \times 10^{-3}$.¹⁰ In ref 11, using NPD at 20 K, the thermal parameters for all the sites were fixed, and esd's for O coordinates were $2\text{--}9 \times 10^{-3}$. These esd's of O coordinates are very large for NPD data. In the structural analysis of Bi₂NiMnO₆ from high-quality synchrotron XRD, the authors had to refine the common thermal parameters for all Bi sites, all Ni/Mn sites, and all O sites.³ The esd's of Bi coordinates were $1\text{--}12 \times 10^{-3}$, which are extremely large for heavy Bi atoms.

All of the above facts and structural information obtained for BiScO₃ in this work indicate that the structural model for BiMnO₃

and related compounds should be revised. There is good indication that the atomic arrangements in the bulk BiMnO₃ are very close to centrosymmetric, if not centrosymmetric at all. Therefore, it seems that the structural distortion cannot be the origin of ferroelectricity in bulk BiMnO₃ as generally accepted. In ref 6, it was noted that noncentrosymmetric distortion in BiMnO₃ is most unusual for the *d*⁴ Mn³⁺ Jahn–Teller ion. In refs 7 and 12, it was suggested that ferroelectricity in BiMnO₃ may originate from an electronic phase transition (electronic ferroelectricity)¹⁹ in contrast to the conventional displacive ferroelectricity caused by a lattice distortion. In ref 20, a very small polarization was obtained for BiMnO₃ in the *C2* model from first-principles calculations. BiScO₃ may serve as a model system to understand a role of Mn³⁺ ions in ferroelectricity of BiMnO₃. We hope that the results of this work will stimulate further research activities in studying the intriguing question about the origin of ferroelectricity in BiMnO₃.

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Supporting Information Available: Details of experiments; selected interatomic distances in the *C2/c* model; observed, calculated, and difference NPD patterns and structure parameters for BiScO₃ refined in space groups *C2* and *Cc* (PDF). This material is available free of charge via the Internet at <http://pubs.acs.org>.

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